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ANSWERS ?HCl aq ?

STRONG ?HC 2 H 3 O 2 aq

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Strong acids are
substances that
completely Strong bases
are substances that
bear a'

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one that is completely
ionized in a solution
Strong Acids
Hydrochloric Acid HCl
Hydrobromic Acid HBr
Sulfuric Acid H₂SO₄'

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Weak Acids What makes a
strong acid strong Why
Acids are substances
that surround us in our
everyday life 2 POGIL'

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Equilibria***

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calculation of H₃O⁺ and pH for solutions of strong acids and

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ANSWERS ?HCl aq ?

STRONG ?HC₂H₃O₂ aq

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answers to differences
between strong and weak
acids and'

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conjugate base of a

strong acid has

little' 'Answer Key

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degrees of dissociation

of strong and weak

acids The photoelectron

spectrum is a plot of

the number of electrons

emitted versus Hold

down the shift'

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calculation of H_3O^+
and pH for solutions of
strong acids and base'

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Weak Acids What makes a
strong acid strong Why*

*Acids are substances
that surround us in our
everyday life 2 POGIL'*

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strong 2 POGIL ?

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