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compound contain 1 mole of C and 2 moles of O'

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air ii 1 mole of carbon is burnt in 16 g of dioxygen'
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class first and adding up kg in kg for weight and liter in
liter for liquids same way the number of atoms molecules
are so large n number because of their very small size so
to signify the count and make sense o'

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June 17th, 2018 - One mole is amount of substance that can be defined as amount of substance containing same number of entities as that in 12g of ^{12}C isotope'

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solution containing 0.2 mole H₂SO₄''Reaction

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Solutions 1 If atomic mass of Mg atom is 24 g find mass of

1 Mg atom Solution We can solve this problem in two ways 1st

way 6.02 × 10²³ amu is 1 g 24''Mole Concept Class 11 XI

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amount of substance which contains equal number of

particles''Mole Concept CBSE Class 11 science

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'Mole Concept Concept of mole molecules mole concept

June 16th, 2018 - Learn about the basics of mole concept

Important questions for class 11 Chemistry In this section

we shall learn about mole concept 1 mole = 6.022 × 10

23''Mole Concept The Basics of the Beginning of Chemistry

June 23rd, 2018 - The mole concept is one of the topics

with which you kick off your preparation of physical

chemistry in class 11 One of the reasons for it being

taught earlier is that the concept of mole will be required

in almost every other topic of physical chemistry that you

study later irrespective of the complexity of that

topic''Chemistry IIT JEE Mains Study Material for Mole

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Mole Concept 8 Amedeo Avogadro found that one gram atom of

any element contains the same number of atoms and one'

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the mole A certain nut crunch cereal contains 11.031

Formula Mass and the Mole Concept by Rice University

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June 10th, 2018 - This lesson is about Mole concept and very helpful to study this confusing topic in simple way'

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