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April 23rd, 2018 - I just finished a lab experiment using copper gluconate as a way to plate copper on to different metals I used a 1M copper gluconate $C_{12}H_{22}CuO_{14}$ I then took a 1 inch strip of aluminum Al zinc Zn iron Fe and tin Sn and submerged them each in a separate test tube with 15 mL of the copper gluconate solution''**Oxidation and Reduction Reactions Redox Reactions**

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Practice Redox Test Answers For questions

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